

# ELIZADE UNIVERSITY, ILARA-MOKIN, ONDO STATE, NIGERIA

DEPARTMENT OF

## MECHANICAL, AUTOMOTIVE AND PRODUCTION ENGINEERING

#### FIRST SEMESTER EXAMINATIONS

#### 2017/2018 ACADEMIC SESSION

**COURSE:** 

GNE 231 – Materials Science (3 Units)

**CLASS:** 

200 Level General Engineering

TIME ALLOWED: 2 Hours: 30 Min.

**INSTRUCTIONS:** Answer any **FOUR** questions

HOD'S SIGNATURE

Date: March, 2018

## Question 1

- a) State Hooke's law and specify the condition under which it is valid (4 marks)
- b) Explain elastic and plastic deformation (4 marks)
- c) Explain using stress/strain curve, the stages metal subjected to load passes through before fracture. (7 marks)

#### Question 2

- a) Explain corrosion and its consequences (4 marks)
- b) Distinguish between oxidation and reduction in electrochemical reaction (4 marks)
- c) Enumerate and explain briefly, five measures that are used to prevent corrosion. (7 marks)

## **Question 3**

- a) Describe the mechanism of crack propagation for both ductile and brittle modes of fracture (7 marks)
- b) Define fatigue and specify the condition under which it occurs (4 marks)
- c) Define creep and specify the condition under which it occurs (4 marks)

#### Question 4

- a) Explain Hot working, Cold working and Heat treatment. (6 marks)
- b) Explain the stages of heat treatment using the temperature/time graph (3 marks)
- c) Explain the different types of heat treatment (6 marks)

## **Question 5**

- a) Explain crystal defect and give the detailed classification (5 marks)
- b) (i) Explain Burger's vector
  - (ii) State the differences between edge dislocation and screw dislocation. (6 marks)
- c) Name four different types of steels and, for each, cite compositional differences among them (4 marks)

## **Question 6**

- a) Derive the relationships between unit cell edge length and atomic radius for body-centered and calculate the atomic packing factor for face centered cubic cubic crystal structures.
  (8 marks)
- b) Explain and draw the unit cells for the principal crystal structures (5 marks)
- c) Describe the difference between crystalline and non-crystalline materials. (2 marks)